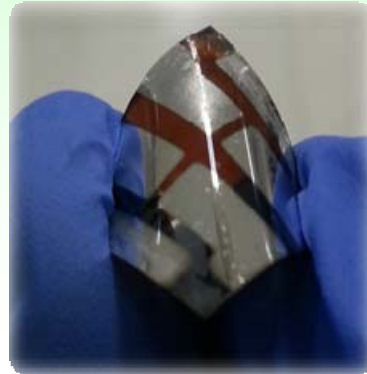


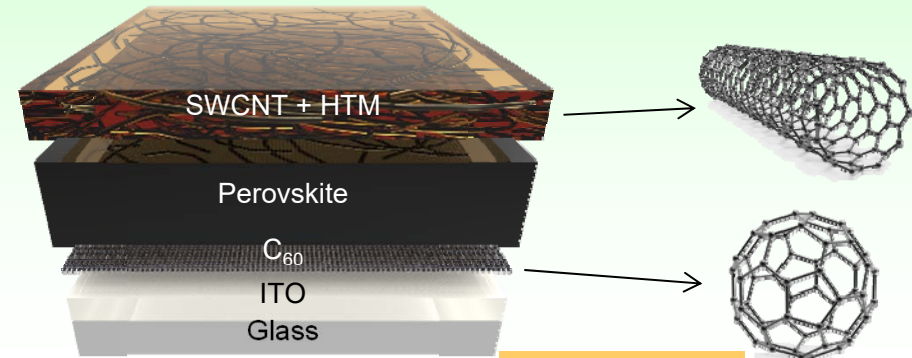
ナノカーボン材料を用いたペロブスカイト型太陽電池



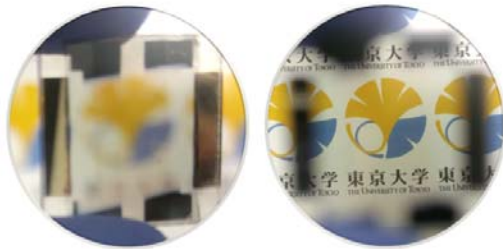
Organic Thin Film
(Normal)



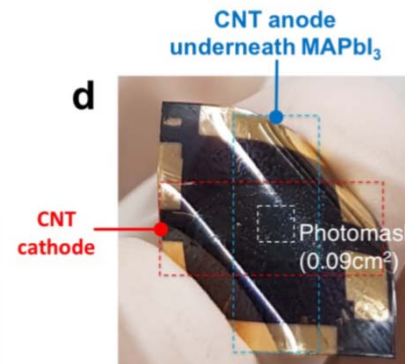
Perovskite
(Inverted)



Perovskite
(Normal)



Organic Thin Film
(Inverted)



Perovskite
CNT for Anode
and Cathode

Shigeo Maruyama (丸山 茂夫)

Department of Mechanical Engineering, The University of Tokyo

Energy NanoEngineering Lab., AIST

College of Chemistry and Molecular Engineering, Peking University

Acknowledgements

日本語
English

Maruyama - Chiashi Laboratory

The University of Tokyo, Department of Mechanical Engineering

Energy NanoEngineering Lab



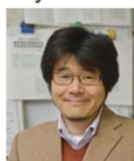
[Japanese Mode is here.](#)

You are the **001449**th visitor, since August 21, 2015

- Energy NanoEngineering Lab. Established on April 1, 2015
- Research Institute of Energy Conservation
- Department of Energy and Environment
- [National Institute of Advanced Industrial Science and Technology \(AIST\)](#)
- 1-2-1 Namiki, Tsukuba, 305-8564, Japan
- [AIST Tsukuba East](#), Main Building (East-1B) 3F, Room 3103
- TEL:029-861-2073 (AIST internal:222-32073)

Members of Maruyama-Chiashi Laboratory

Updated: May 1, 2017
Faculty & Staff



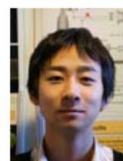
Professor:
[Shigeo Maruyama](#)
[maruyama]



Associate Prof.:
[Shohei Chiashi](#)
[chiashi]



Assistant Prof.:
[Xiang Rong](#)
[xiangrong]



Assistant Prof.:
[Taiki Inoue](#)
[inoue]



Visiting Prof.:
[Kazutomo Suenaga](#)
[suenaga-kazu]



Visiting Prof.:
[Yan Li](#)
[yanli]

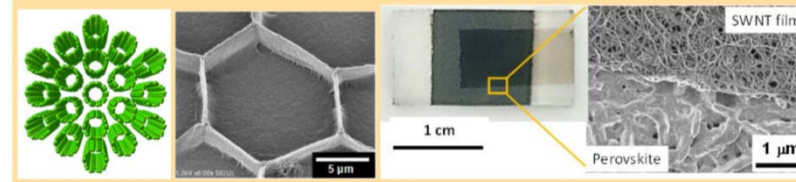


Visiting Prof.:
[Yutaka Matsuo](#)
[matsuo]



Project Assistant
Prof.:
[Jeon Il](#)
[il.jeon]

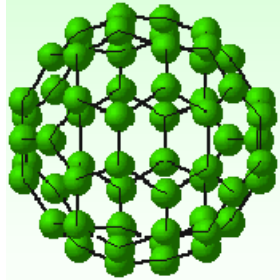
Research Direction:



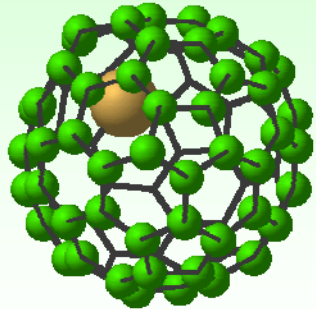
April 20, 2016



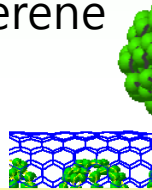
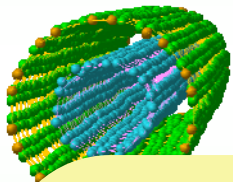
Research Society



Fullerene



Metallofullerene



FT

1991: C60 Research Associ

1991: *President: Professor*

1995: The Fullerenes Resea

2002: The Fullerene Nanota

2003: *President: Professor Hu*

2005: The Fullerenes and Nanotubes research society

2011: The Fullerenes, Nanotubes and Graphene Research Society

2011: *President: Professor Shigeo Maruyama (UTokyo)*

FNTG Research Society

Meetings

2017/3/1-3: FNTG 52 Symp. @ Tokyo

2017/6/25-30: NT17 @ Brazil

2017/9/13-15: FNTG 53 Symp. @ Kyoto

2018/3/10-12: FNTG 54 Symp. @Tokyo

2018/7/8-12: WONTON2018 @ Hakone

2018/7/15-20: NT18 @ Beijing

2018/9/11-13: FNTG 55 Symp. @ Sendai

2019/2/??-??: FNTG 56 Symp. @ Tokyo

2019/7/21-26: NT19 @ Würzburg

2019/9/?-?: FNTG 57 Symp. @ Nagoya

...ond

IRENA

Indium Replacement by Single-Walled Carbon Nanotube Thin Films



S. Maruyama



E. I. Kauppinen



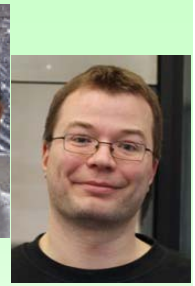
H. Shinohara



C. Bichara



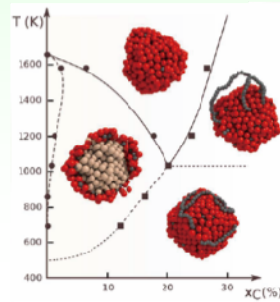
Y. Ohno



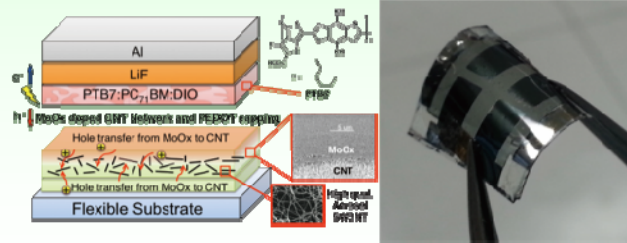
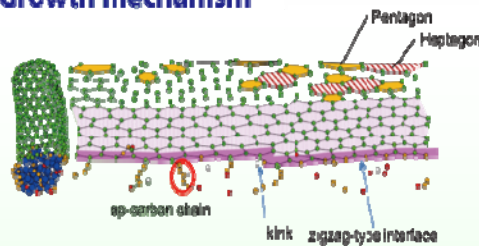
J. Wagner

Simulation techniques

Role of catalyst



Growth mechanism

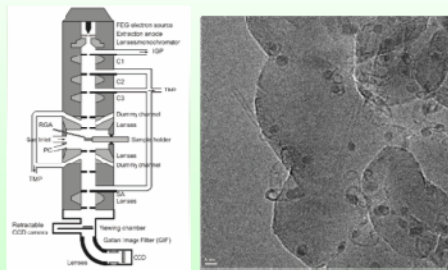


One-step transfer printing fabrication of touch sensors

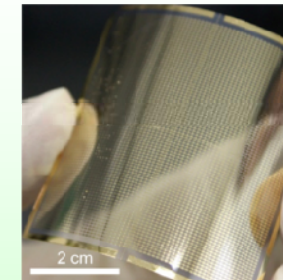


State-of-the-art CNT Films & Device Technologies for Flex, 3D Electronics

In-situ TEM analysis



Large scale, rapid separation



High-performance Flex TFTs

Partners:

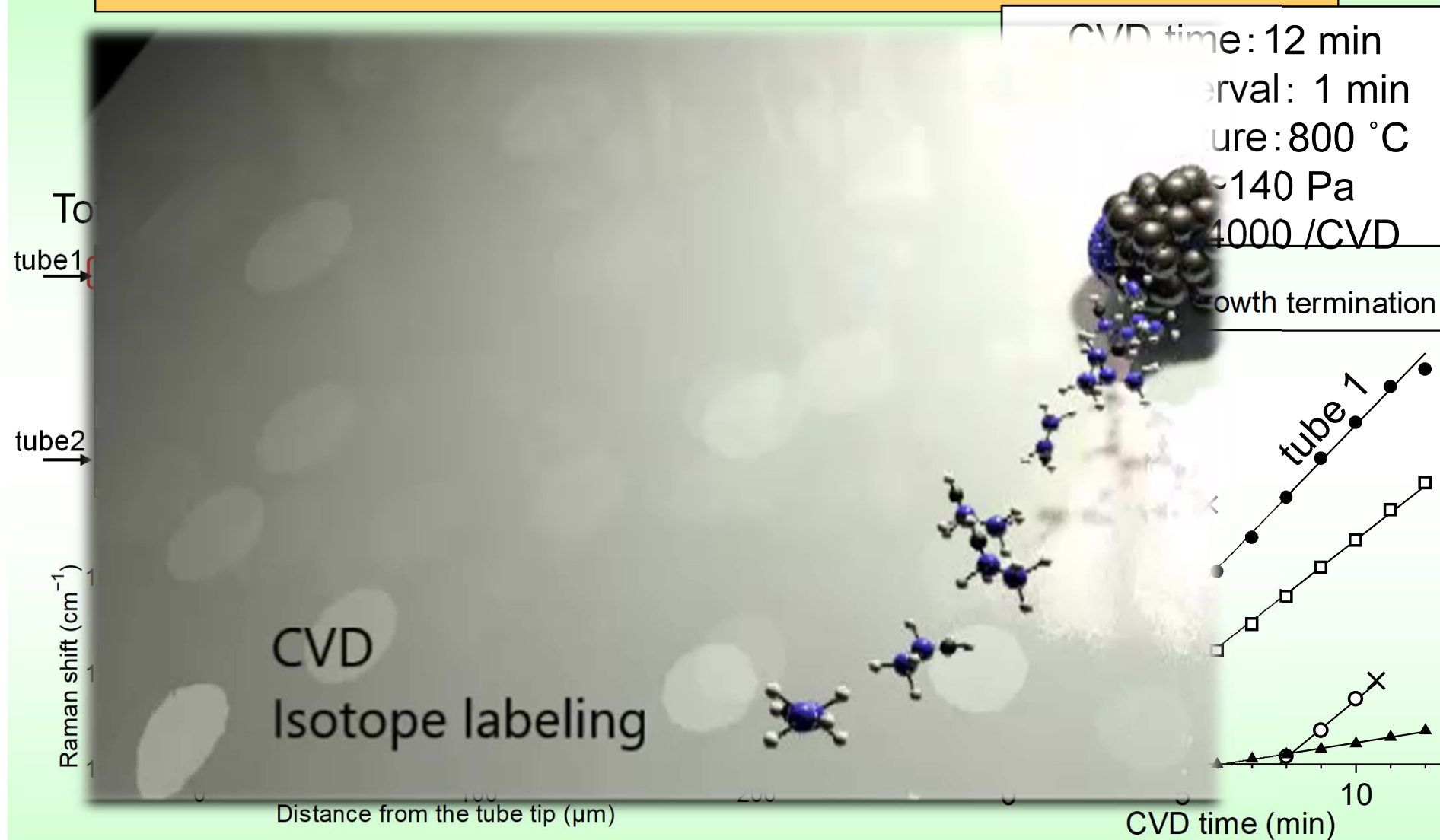


Technical University of Denmark



THE UNIVERSITY OF TOKYO

Time evolution of SWNT lengths

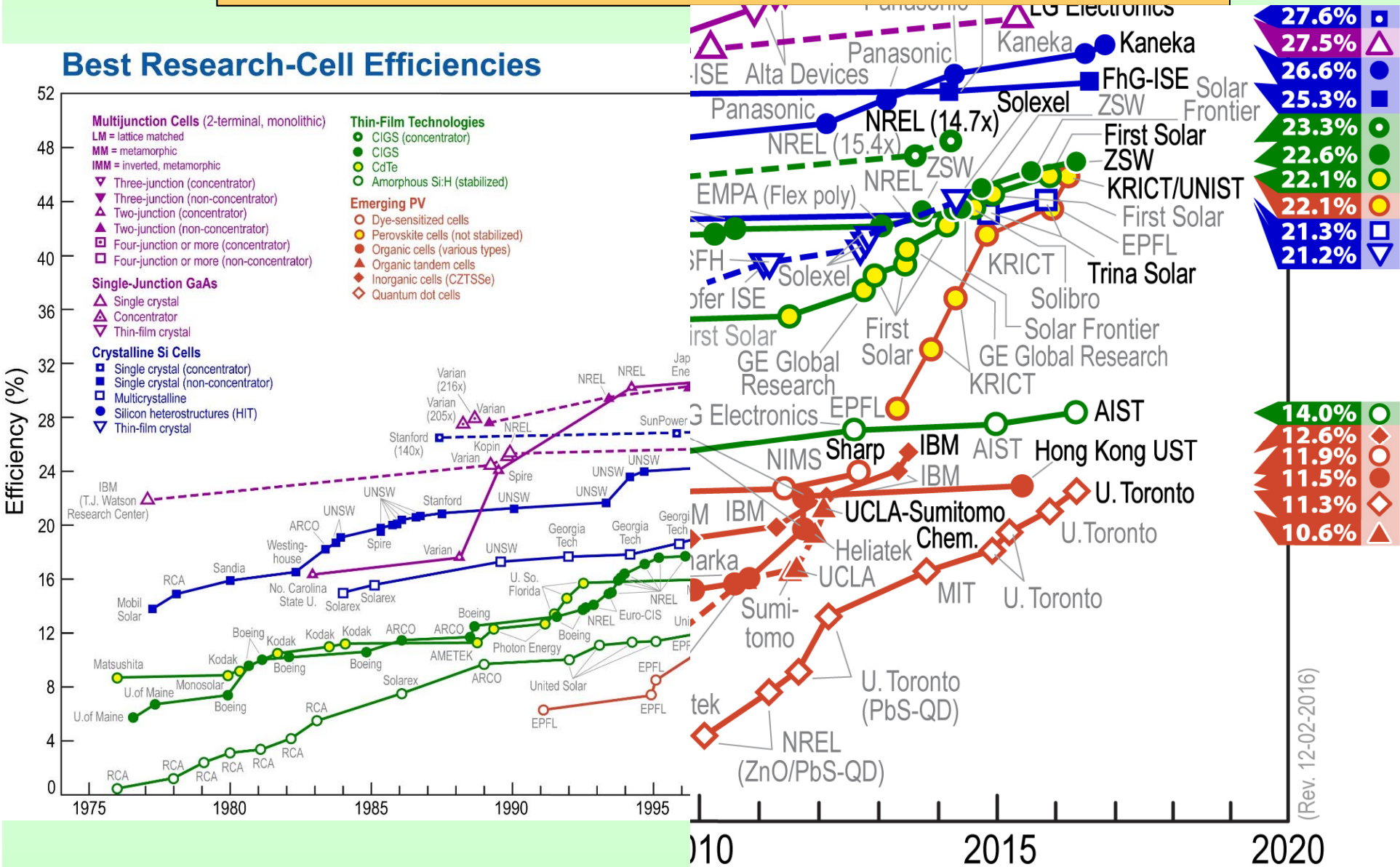


➤ Sudden stop of growth ($\Leftrightarrow L = \gamma T_0(\exp(-t/T_0))$ [1])

➤ Different K. Otsuka et al., ACS Nano (2018) [DOI: 10.1021/acsnano.8b01630]

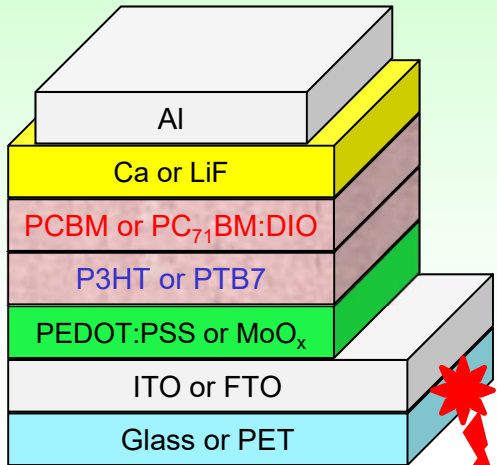
NREL solar cell efficiency chart

Best Research-Cell Efficiencies

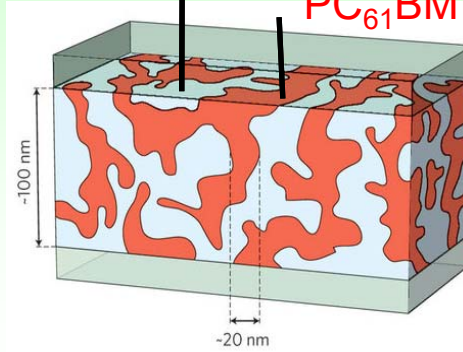


Organic Thin Film

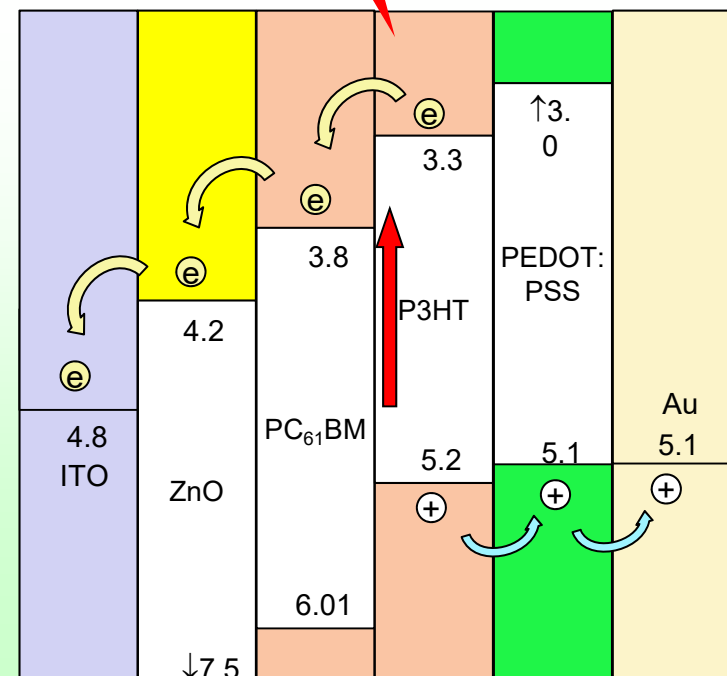
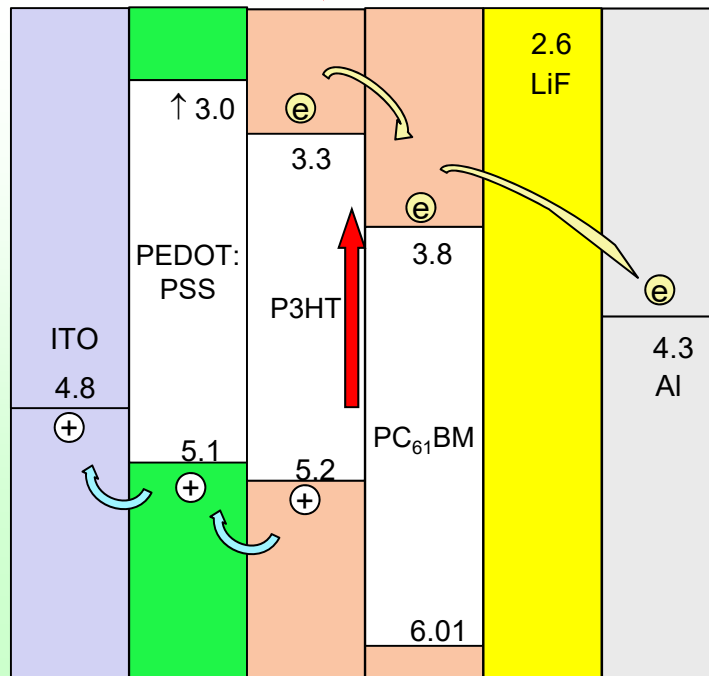
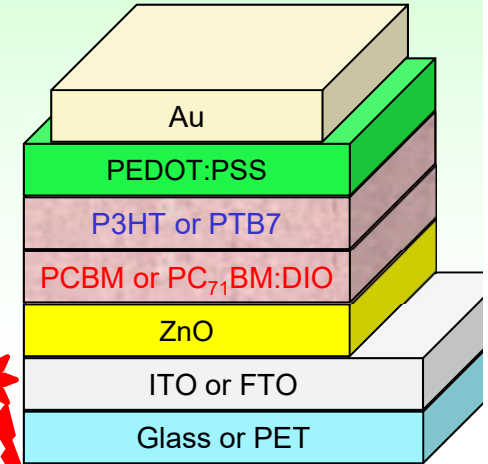
Normal structure



P3HT Donor
PC₆₁BM Acceptor

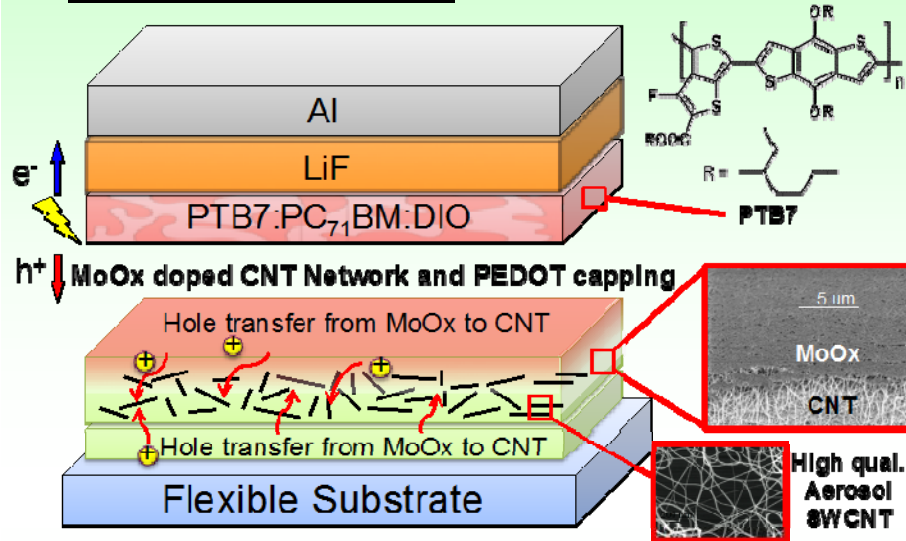


Inverted structure

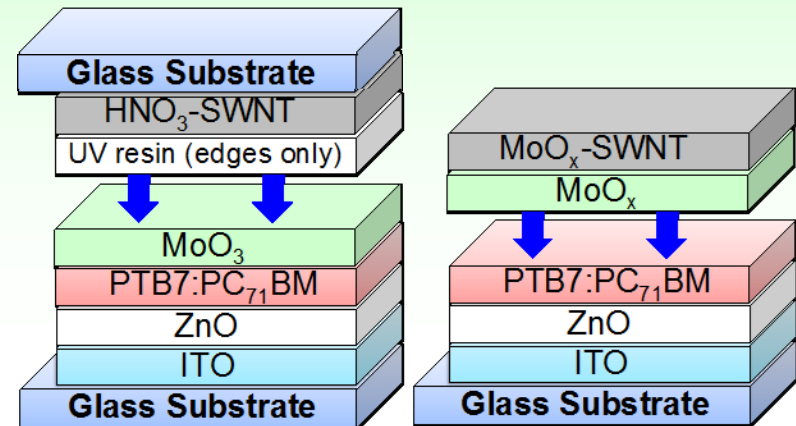


CNT for Organic Thin Film

Normal structure



Inverted structure

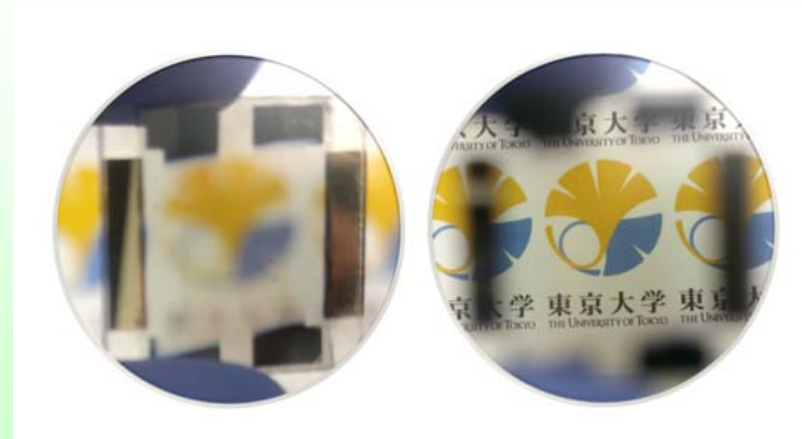
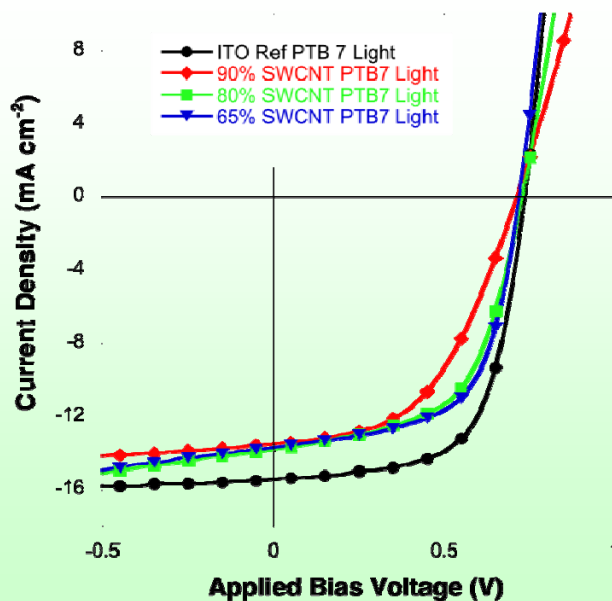


Sandwich HNO₃:3.7 %

Bridge MoO_x:3.1 %

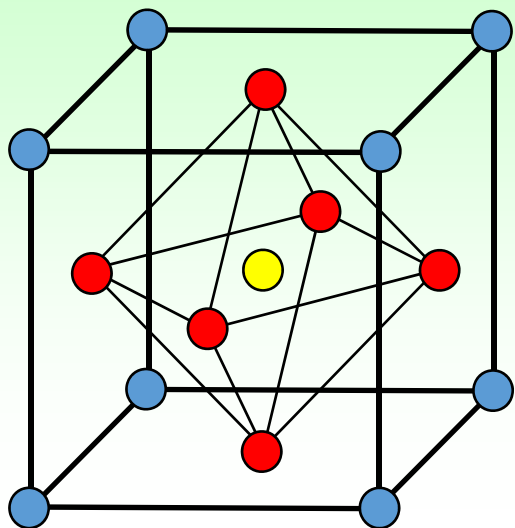


Il Jeon et al.,
JACS
137 (2015) 7982.



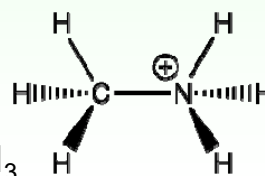
I. Jeon et al., Sci. Rep., 6 (2016) 31348.

Perovskite Solar Cells



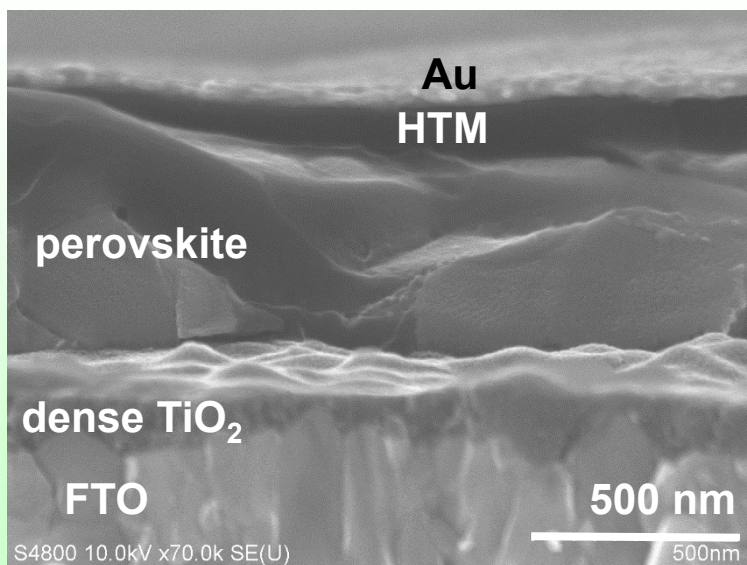
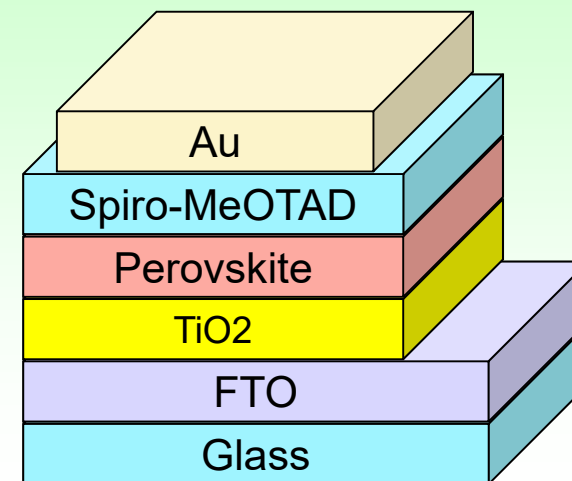
Perovskite: AMX_3

- A = Methylammonium (MA), $CH_3NH_3^+$
Formamidinium (FA), $CH_3(NH_2)_2^+$
Guanidinium (Gua), $CH_6N_3^+$
- M = Pb^{2+} , Sn^{2+} , Cs^{2+} ,
 Rb^{2+}
- X = I, Cl, Br

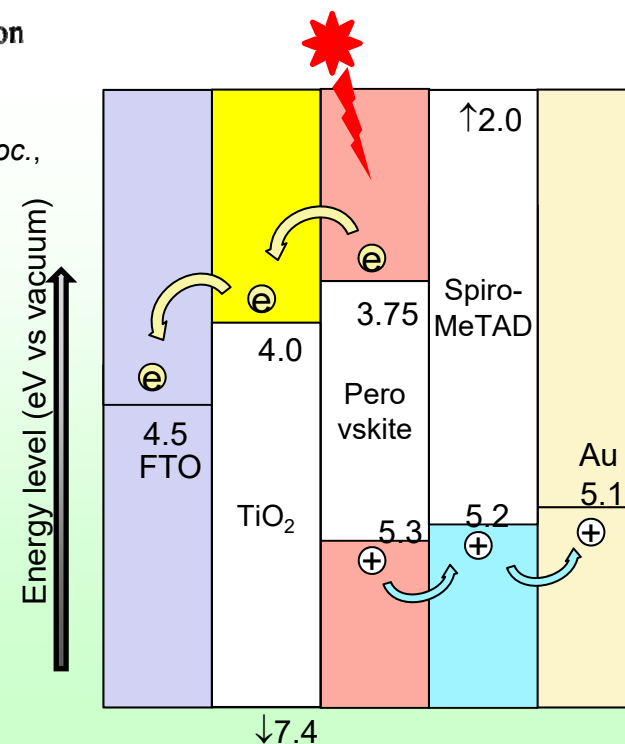


MAPbI₃: $CH_3NH_3PbI_3$

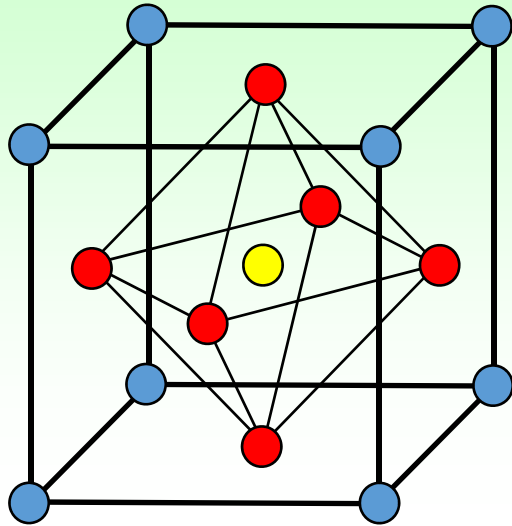
methylammonium ion



- Use of Perovskite **PCE 3.8 %**
A. Kojima, *et al.*, *J. Am. Chem. Soc.*, **2009**
- Solid state solar cell **PCE 10.9 %**
M. M. Lee, *et al.*, *Science*, **2012**
- High **PCE 20.1 %**
N. J. Jeon, *et al.*, *Nature*, **2014**
- **PCE record 22.1% 2016**



Perovskite Solar Cells

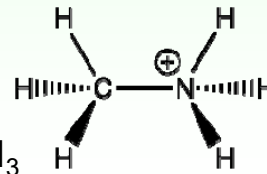


Perovskite: AMX_3

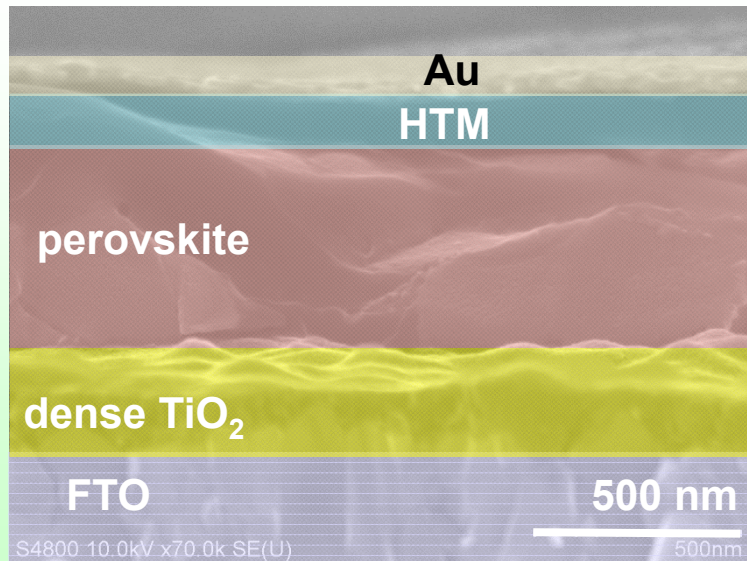
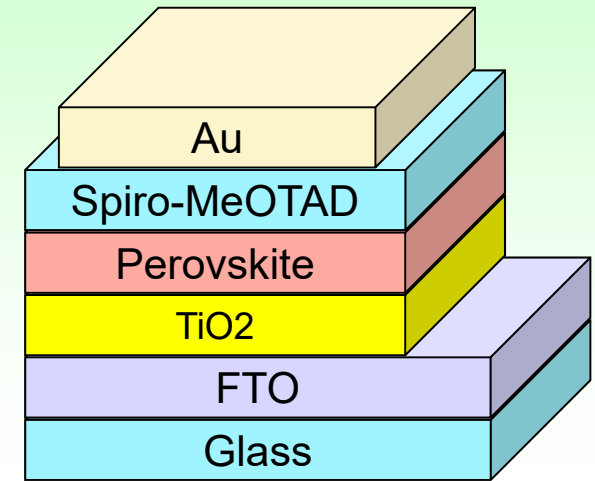
● A = Methylammonium (MA), $CH_3NH_3^+$
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 Guanidinium (Gua), $CH_6N_3^+$

● M = Pb^{2+} , Sn^{2+} , Cs^{2+} ,
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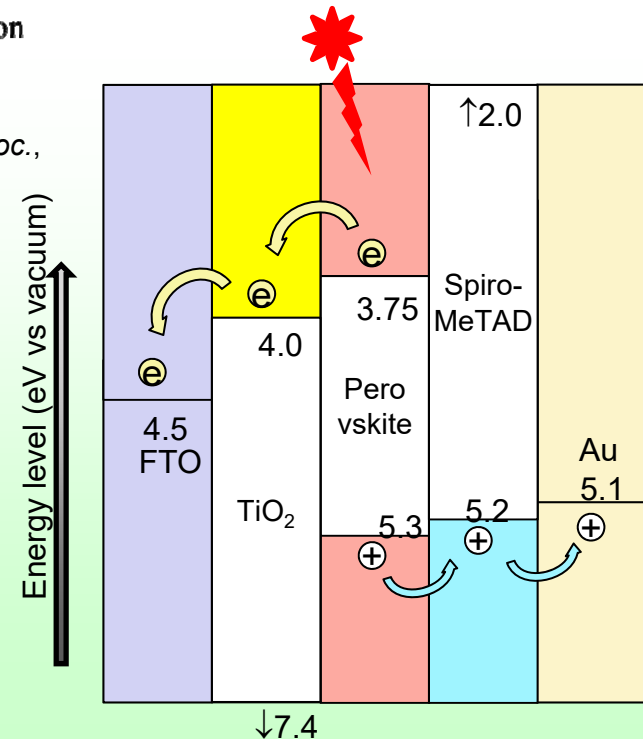
● X = I, Cl, Br



methylammonium ion

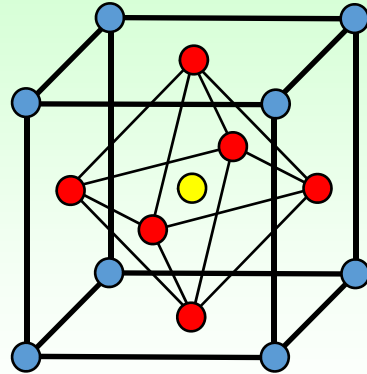
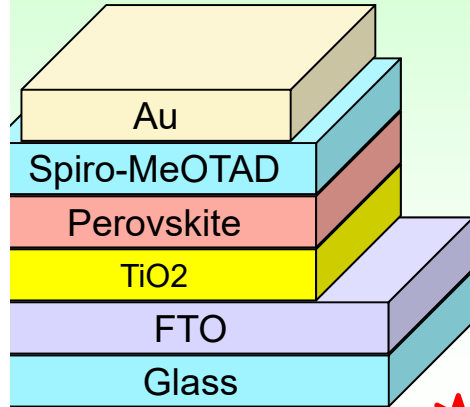


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 N. J. Jeon, *et al.*, *Nature*, **2014**
- **PCE record 22.1% 2016**



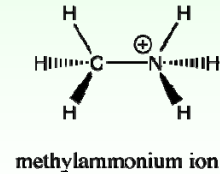
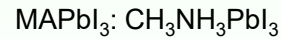
Organic-Inorganic Hybrid Perovskite Solar Cells

Normal structure

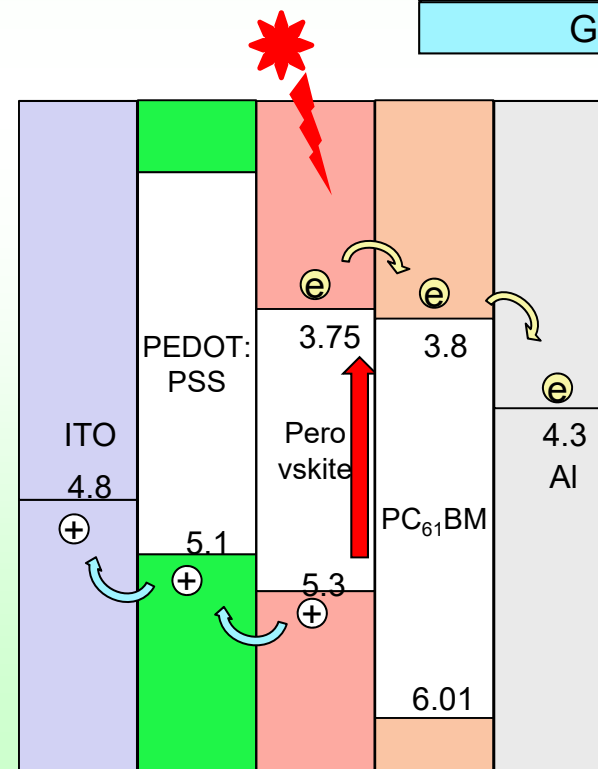
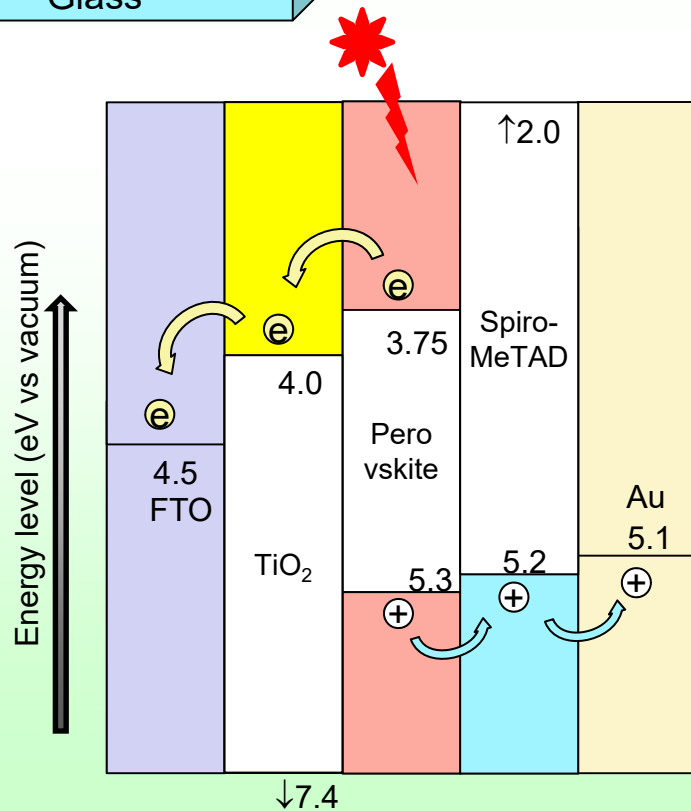
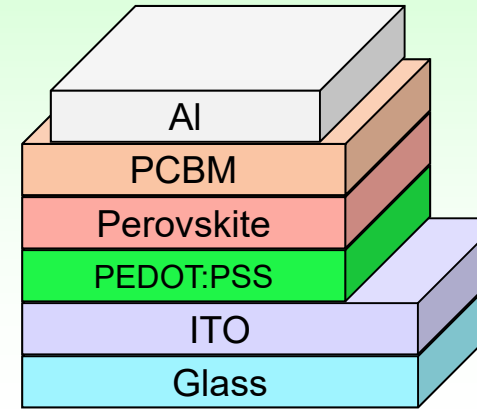


Perovskite: AMX_3

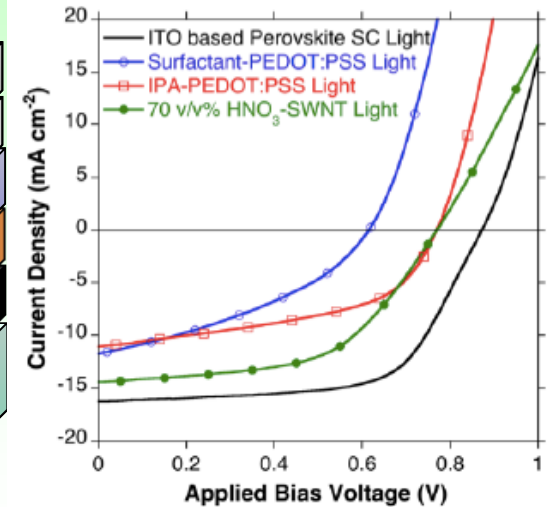
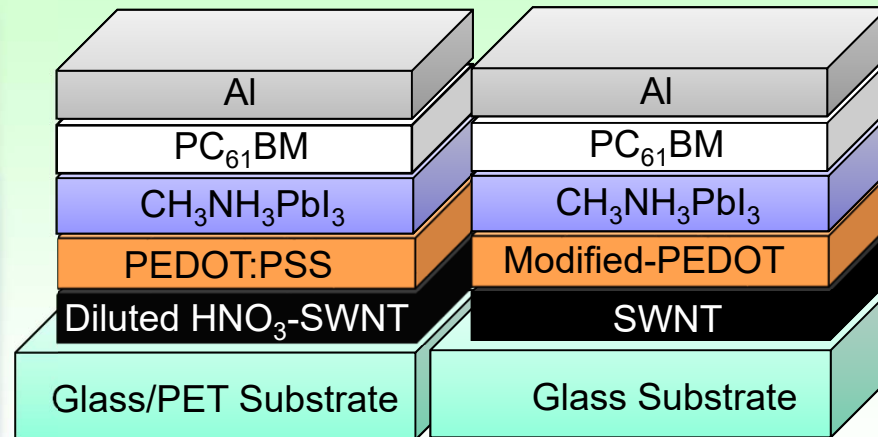
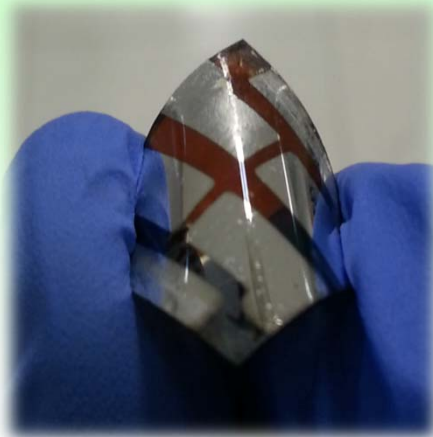
- A = Methylammonium (MA), $CH_3NH_3^+$
Formamidinium (FA), $CH_3(NH_2)_2^+$
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 Rb^{2+}
- X = I, Cl, Br



Inverted structure



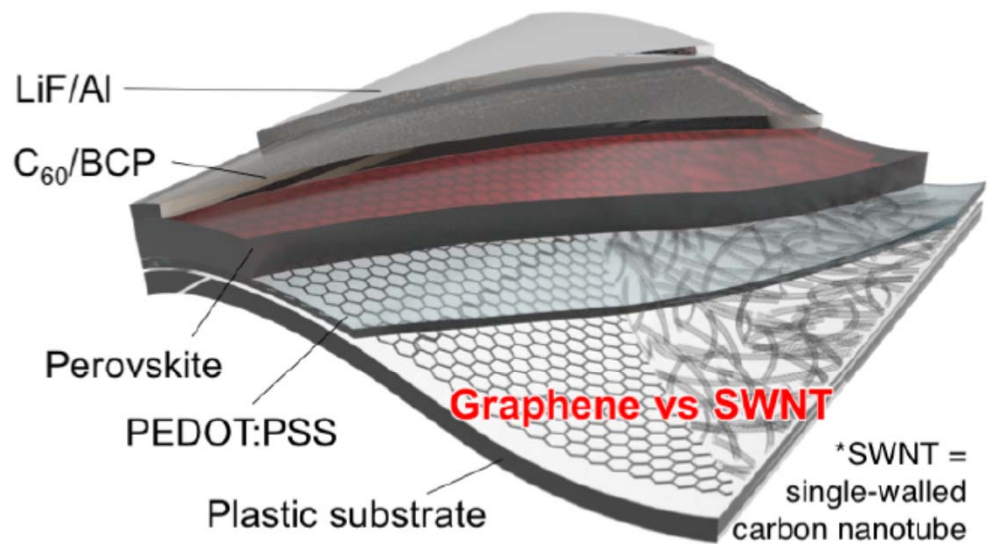
ITO-Free Perovskite Solar Cells



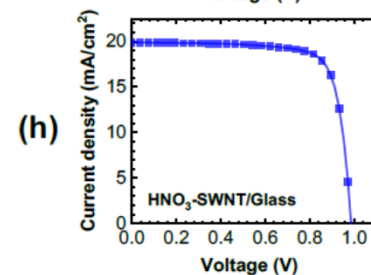
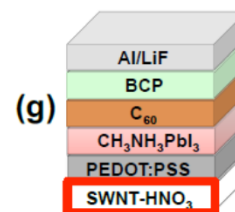
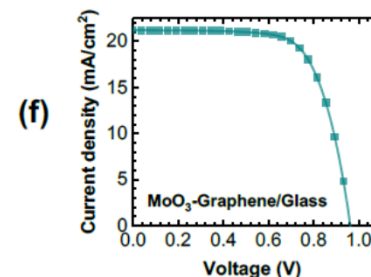
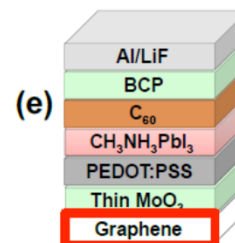
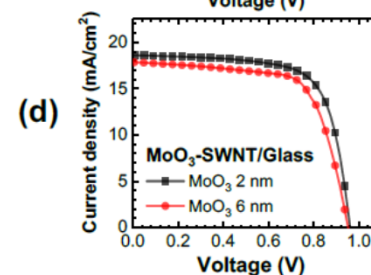
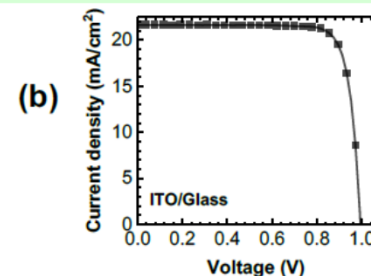
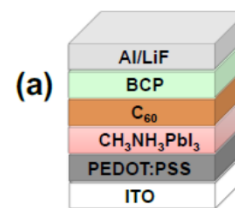
Substrate	Anode	HTL	V_{oc} (V)	J_{sc} (mA/cm ²)	FF	PCE (%)
Glass	ITO	PEDOT:PSS	0.83	16.3	0.64	9.05
Glass	SWNT	IPA-PEDOT:PSS	0.77	11.1	0.50	4.27
Glass	SWNT	Surfactant-PEDOT:PSS	0.61	11.8	0.38	2.71
Glass	70 v/v% HNO ₃ -SWNT	PEDOT:PSS	0.77	14.4	0.55	6.09
Glass	50 v/v% HNO ₃ -SWNT	PEDOT:PSS	0.76	14.5	0.52	5.84
Glass	35 v/v% HNO₃-SWNT	PEDOT:PSS	0.79	14.9	0.54	6.32
Glass	15 v/v% HNO ₃ -SWNT	PEDOT:PSS	0.77	13.6	0.39	3.88
PET	35 v/v% HNO₃-SWNT	PEDOT:PSS	0.71	11.80	0.56	5.38

I. Jeon, T. Chiba, C. Delacou, Y. Guo, A. Kaskela, O. Reynaud, E. I. Kauppinen, S. Maruyama, Y. Matsuo, *Nano Lett.*, 15 (2015) 6665.

Carbon nanotubes vs graphene for inverted PSC

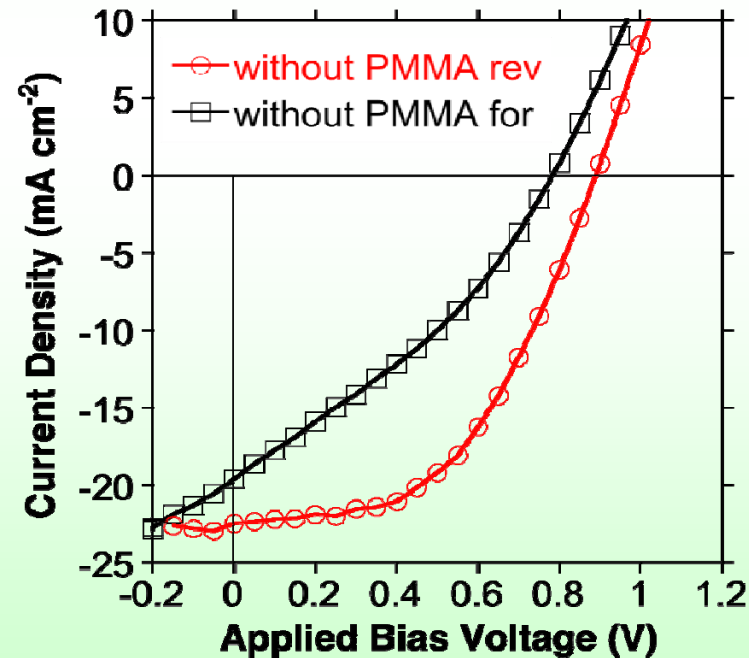
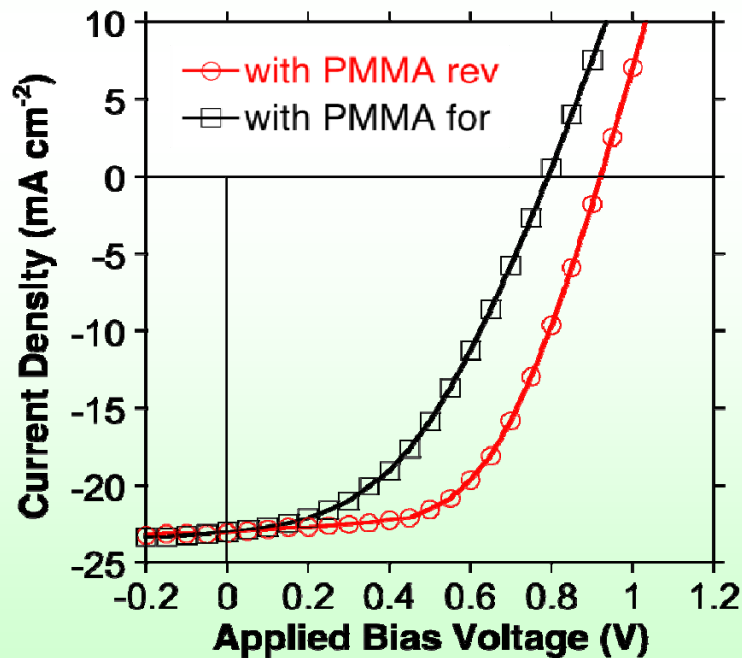
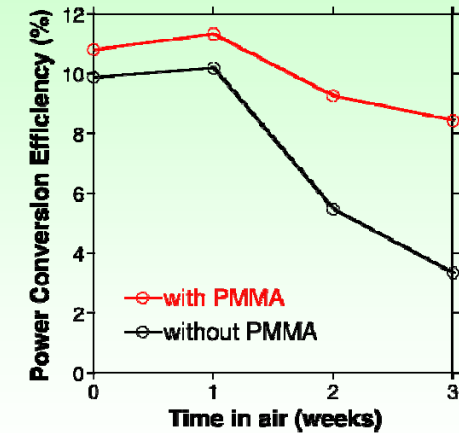
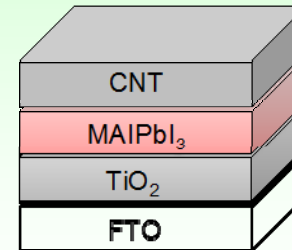
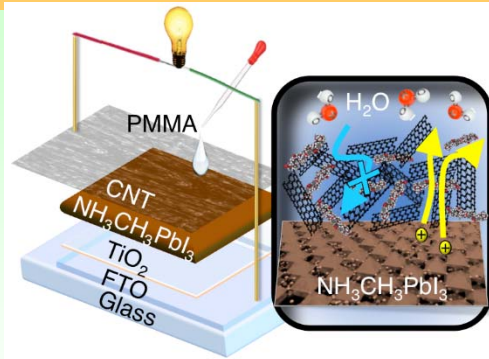
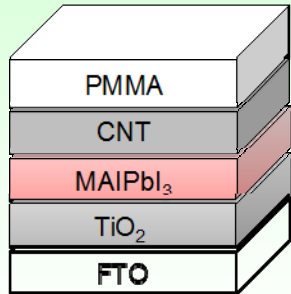


Efficiency	14.2%	12.8%
Reproducibility	moderate	high
Cyclic flexibility	flexible	very flexible



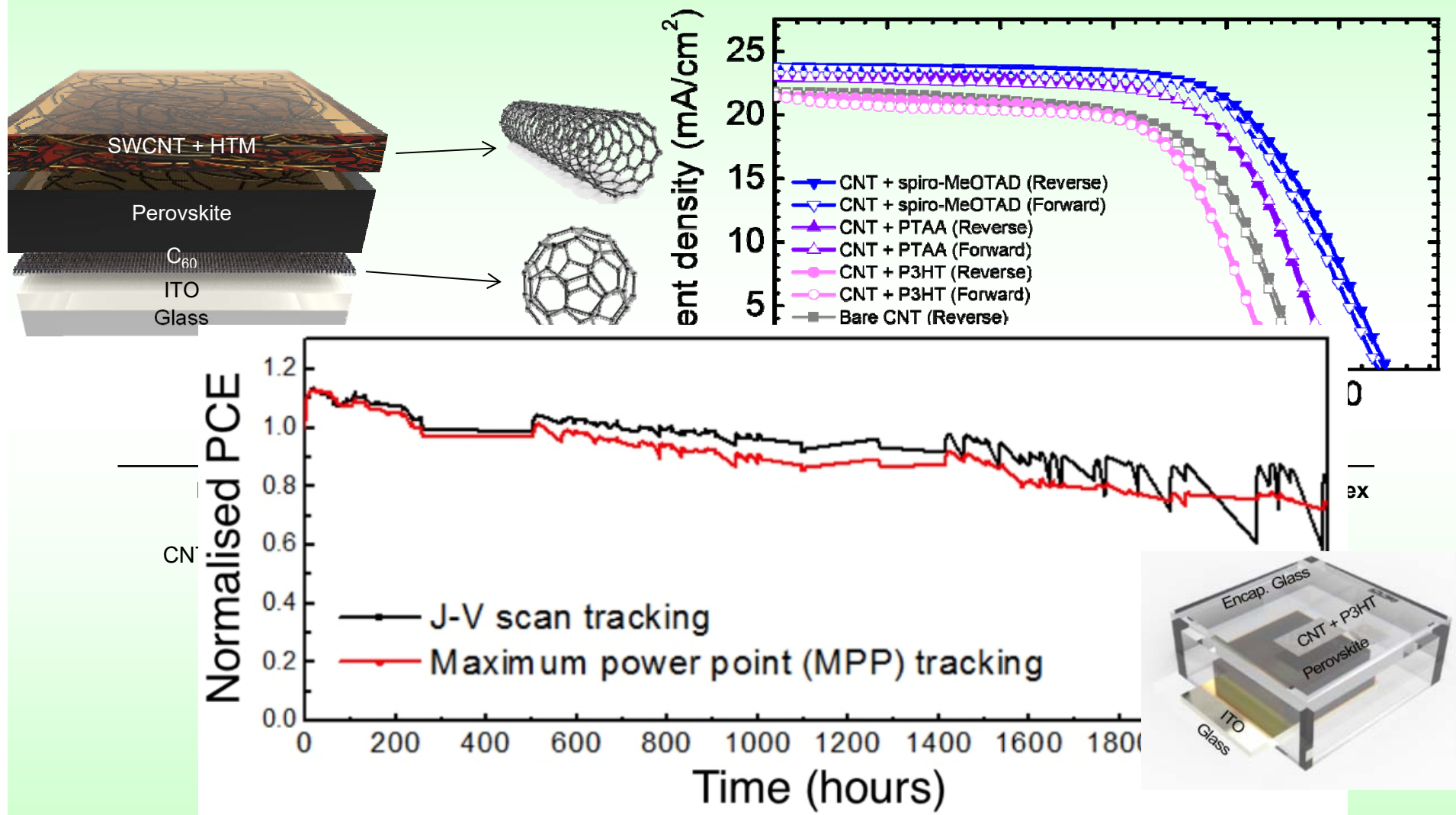
I. Jeon, J. Yoon, N. Ahn, M. Atwa, C. Delacou, A. Anisimov, E. Kauppinen, M. Choi, S. Maruyama, Y. Matsuo, J. Phys. Chem. Lett., 8 (2017) 5395.

CNT Electrode (PMMA Layer for Stability)



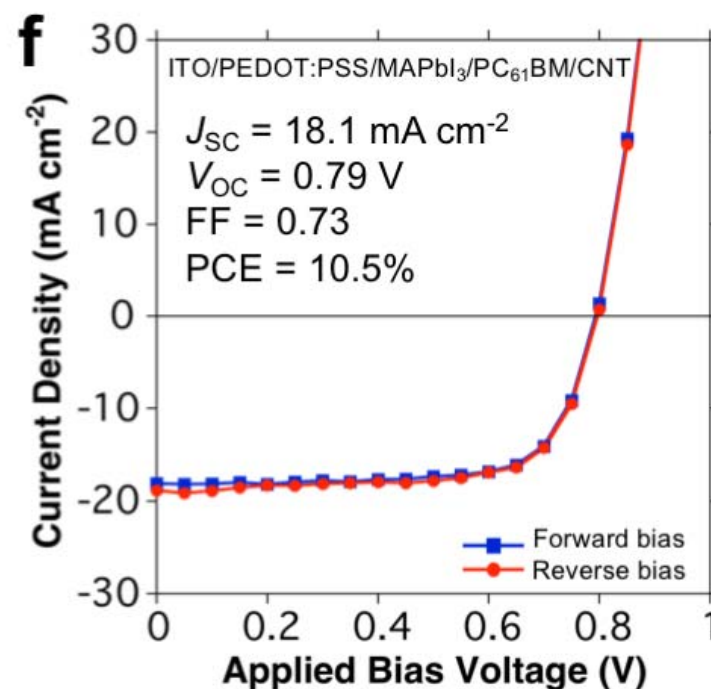
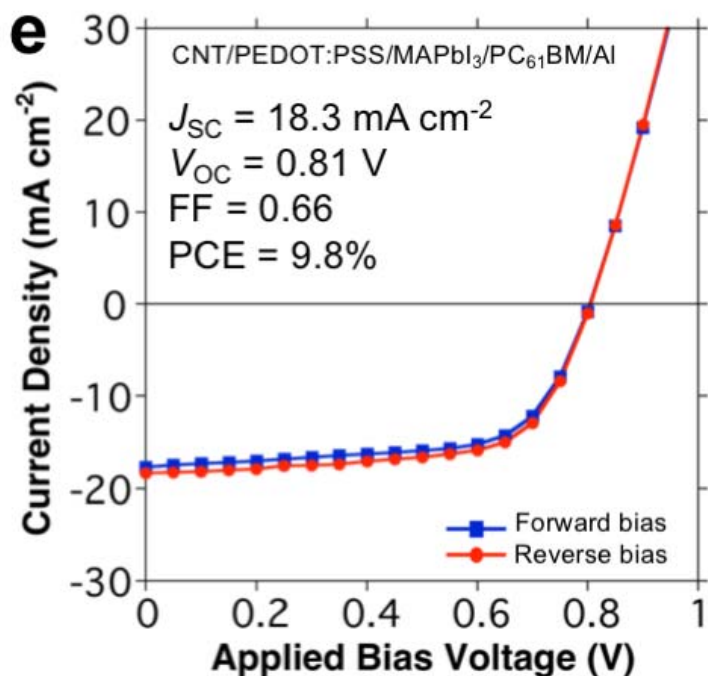
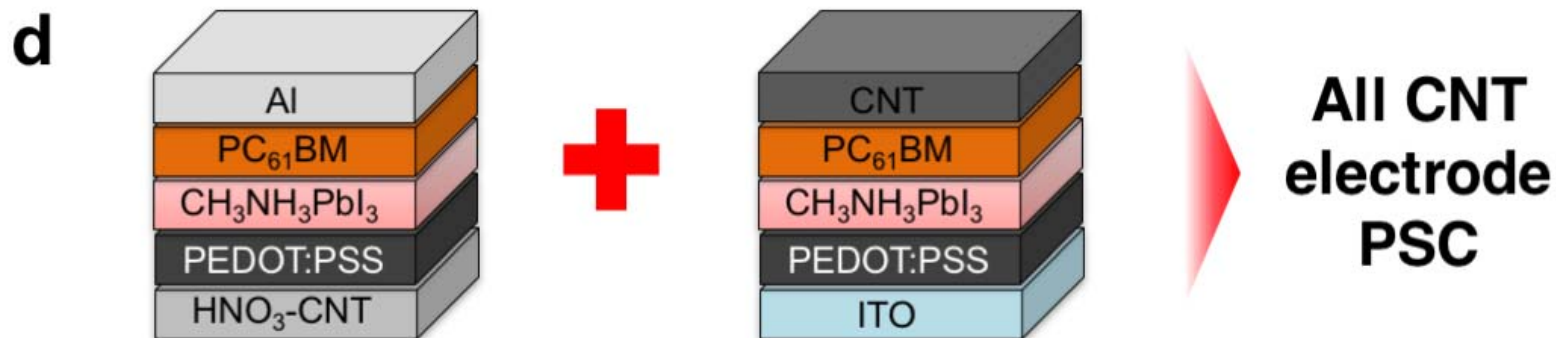
Takahiro Sakaguchi, Il Jeon, Takaaki Chiba, Ahmed Shawky, Esko I. Kauppinen, Rong Xiang, Shohei Chiashi, Nam-Gyu Park*, Yutaka Matsuo*, Shigeo Maruyama*

CNT Electrode in Perovskite Solar Cells

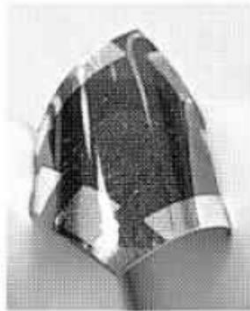


Namyoun Ahn, Il Jeon, Jungjin Yoon, Esko I. Kauppinen, Yutaka Matsuo*,
Shigeo Maruyama*, Mansoo Choi*, J. Mater. Chem. A, 6(2018)1382.

PC₆₁BM-Soaked CNTs as Cathode



Il Jeon, Seungju Seo, Yuta Sato, Clement Delacou, Anton Anisimov, Kazu Suenaga, Esko I. Kauppinen, Shigeo Maruyama*, Yutaka Matsuo*, J. Phys. Chem. C, 121 (2017) 25743.



東京大学の丸山茂夫教授と松尾豊特任教授らは、製造コストを半分にできる薄型太陽電池を開発した。電極材料を従来の金属から筒状炭素分子「カーボンナノチューブ」に変え、電気代などコストがかさむ真空での作業工程をなくした。材

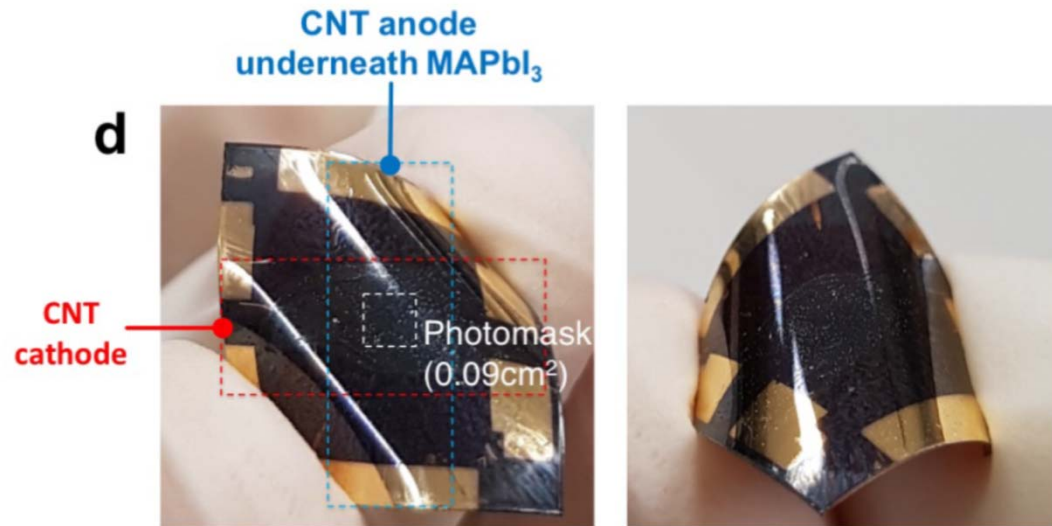
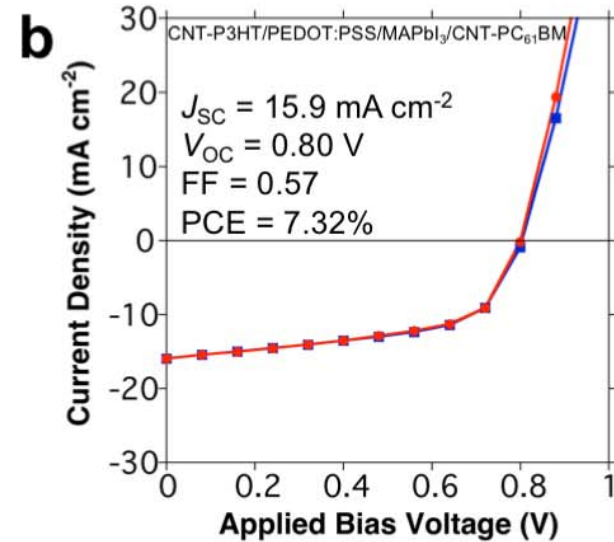
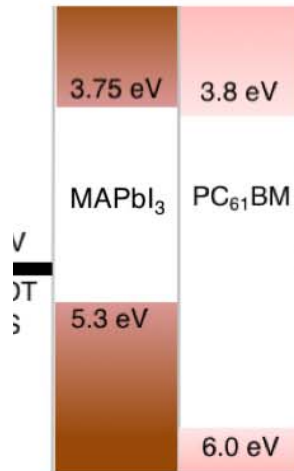
太陽電池 製造コスト半分

素材 金属からナノチューブに

東大、車など曲面向け
 材料メーカーなどと組み、数年後の実用化を目指す。開発したのは「ペロブスカイト型」と呼ばれる有機太陽電池の一種。写真は東大提供。薄く柔らかいので折り曲げて、自動車の車体や建物の曲面などに設置できる。1.5枚角の試作品は電極にインジウムなどの代わりにカーボンナノチューブを使った。球状炭素分子「フラーレン」の一種

と組み合わせて電気を取り出しやすくした。発電効率は7%と主流のシリコン製より低いが、今後の研究で改善できるとみている。金属製電極を作る場合、高価な製造設備を使う場合、蒸着といった工程が必要だった。カーボンナノチューブは合成樹脂などの基板に塗るだけで電極になる。使用量も少なく全体では安く

Cathode and Anode Electrodes

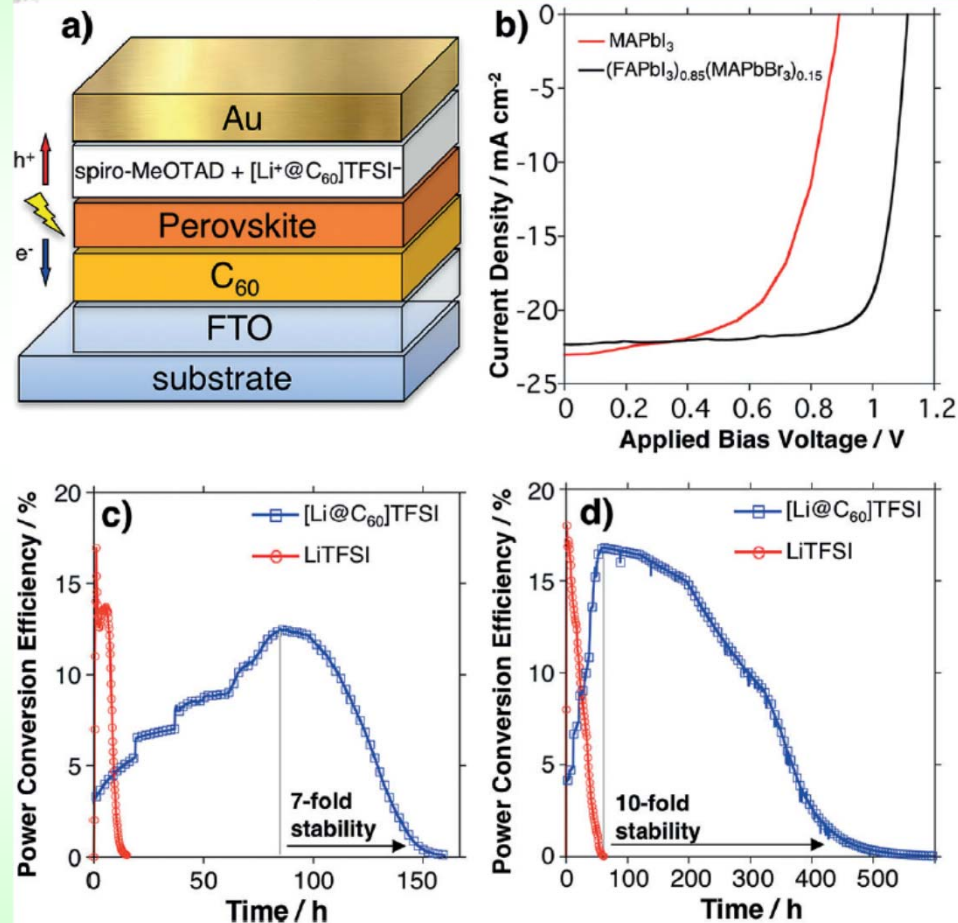
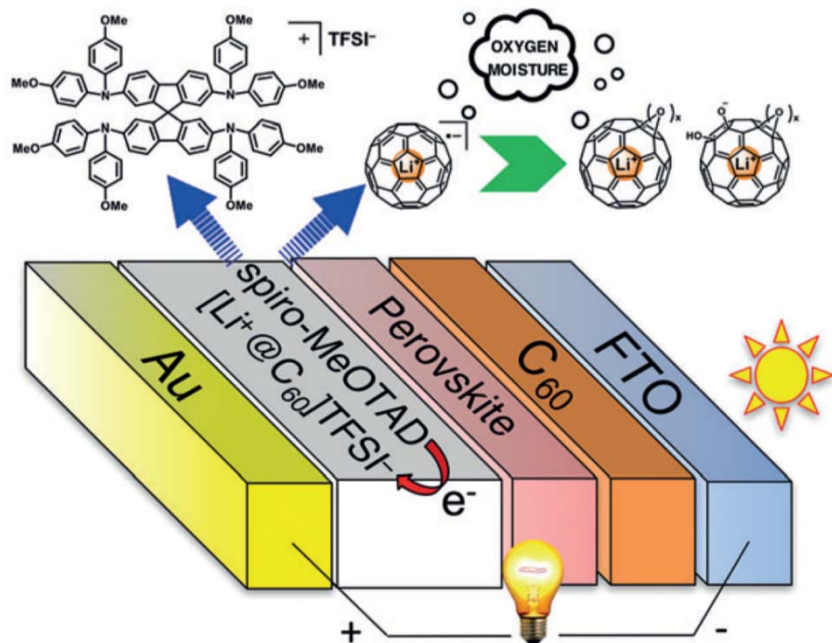


to, Clement Delacou, Anton Anisimov, Kazu Suenaga, ma*, Yutaka Matsuo*, J. Phys. Chem. C, 121 (2017) 25743.

Metal endohedral fullerene, $\text{Li}^+@C_{60}$

Instant Oxidation

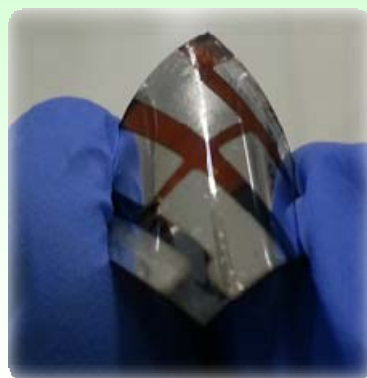
Anti-oxidant activity



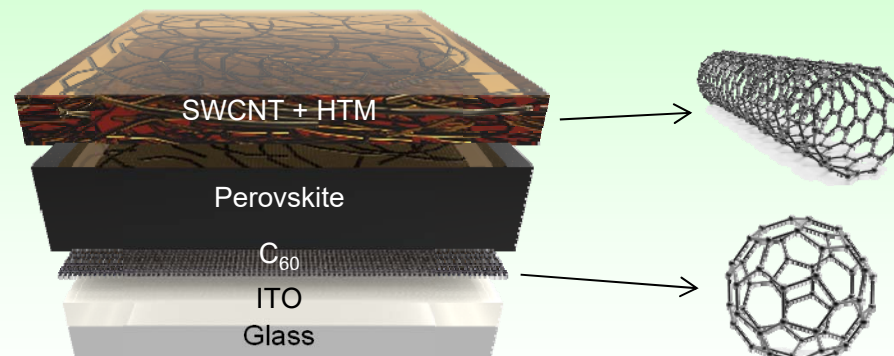
I. Jeon, H. Ueno, S. Seo, K. Aitola, R. Nishikubo, A. Saeki, H. Okada, G. Boschloo, S. Maruyama, Y. Matsuo, *Angew. Chem. Int. Ed.* 2018, in press. [DOI: 10.1002/anie.201800816].



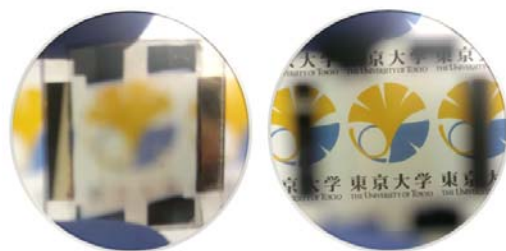
Organic Thin Film
(Normal)



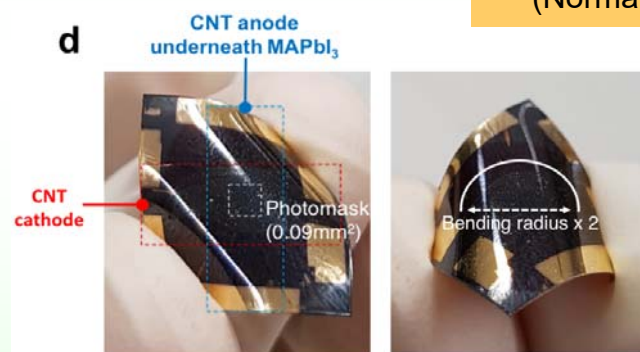
Perovskite
(Inverted)



Perovskite
(Normal)



Organic Thin Film
(Inverted)



Perovskite
CNT for Anode
and Cathode

SATL 
Science of Atomic Layers

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